



JACOBS
UNIVERSITY



Study Program Handbook

Chemistry

Bachelor of Science

Subject-specific Examination Regulations for Chemistry (Fachspezifische Prüfungsordnung)

The subject-specific examination regulations for Chemistry are defined by this program handbook and are valid only in combination with the General Examination Regulations for Undergraduate degree programs (General Examination Regulations = Rahmenprüfungsordnung). This handbook also contains the program-specific Mandatory Module and Examination Plans (Appendix 1a / 1b).

Upon graduation, students in this program will receive a Bachelor of Science (BSc) degree with a scope of 180 ECTS (for specifics see chapter 3 of this handbook).

Version	Valid as of	Decision	Details
Fall 2018 - V1	01.09.18	Academic Senate August 29, 2018	Master Version

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1 The Chemistry Study Program

1.1 Concept

Chemistry relates to nearly every aspect of our lives. It lies at the heart of some of the world's most advanced industries, for example those focusing on pharmaceuticals, sustainable energy development, or even the food and beverage industry. The transdisciplinary Bachelor degree in Chemistry offers modules that include courses in nanomolecular science, organic, inorganic, analytical, and physical chemistry; where students are also taught the relevant aspects of mathematics, engineering, and industrial priorities. Over the course of the three-year study program you will take extensive laboratory courses and conduct your own research projects during your third year of study. Undergraduate students are also strongly encouraged to engage in research projects with graduate students as early as their first or second semester at Jacobs University.

1.2 Specific Advantages of the Chemistry Program at Jacobs University

- Chemistry at Jacobs University combines the long tradition of a thorough education in Chemistry with the opportunity to participate in research projects already at a very early stage of the studies.
- The first year lays the foundation in chemistry, the other natural sciences, and mathematics. The second year focuses on the core education in Organic, Inorganic, and Physical Chemistry. Lectures and exercises furnish theoretical knowledge, while extensive laboratory modules put in place the required practical skills. During the third year, the students carry out their own guided research project in the laboratory of a faculty member in Chemistry.
- Chemistry at Jacobs University provides a stimulating study environment. As classes are small, there is good and supportive interaction of the Chemistry Faculty with the students, with regular contact and support in theoretical questions, research, and the development of written and oral communication skills. Undergraduate students are encouraged, at an early stage, to get involved in a research project that allows close interaction with M.Sc. and Ph.D. students.
- Chemistry at Jacobs University offers a challenging learning environment with maximum flexibility. These unique features combine to make our graduates very attractive to employers as well as graduate schools.

1.3 Program-Specific Qualification Aims

Transferable Skills

Students will be able to

- perform literature searches as structured approaches, and extract peer-reviewed information from the primary literature at large.
- analyze scientific and technical questions, put them into relationship to what is known, and suggest ways to solve them.
- demonstrate a general skill set of scientific methods used in chemistry.

- present own results, and those of others, concisely and professionally in front of a knowledgeable audience.
- use databanks to retrieve chemical information.

Theory

Students will be able to

- understand the history and role of the periodic table of elements and identify trends in groups and periods.
- explain the concepts of chemical bonding, conformation and stereochemistry, the major functional groups and their manipulations, chemical synthesis, and naturally occurring chemical compounds (natural products).
- perform stoichiometric calculations.
- predict the existence and properties of known and unknown compounds and materials, and propose suitable synthesis pathways and characterization techniques.
- explain the thermodynamic and kinetic laws of relevance for chemistry.
- put into context the chemistry, thermodynamics, and kinetics of chemical reactions.
- explain the principles behind representative analytical methods used for identification and quantification of chemical compounds.
- explain how the use of biological systems (enzymes and cells) and renewable raw materials can help to make synthesis of chemicals more sustainable.
- explain weak, noncovalent forces and how they affect the properties of materials and the function of living systems

Practical Work

Students will be able to

- know about safety aspects and the guidelines for good academic practice.
- perform selected chemical syntheses in organic and in inorganic chemistry.
- possess basic synthetic/analytical lab skills and be able to use basic laboratory equipment.
- demonstrate practical insight into the experimental analysis of physicochemical parameters.
- quantify chemical compounds.
- elucidate the structure of chemical compounds from analytical measurements.
- perform binding titrations and use spectroscopic methods for the investigation of intermolecular interactions
- prepare comprehensive lab reports.

1.4 The Jacobs University Employability and Personal Development Concept

Jacobs University's educational concept aims at fostering employability which refers to skills, capacities, and competencies which transcend disciplinary knowledge and allow graduates to quickly adapt to professional contexts. Jacobs University defines employability as encompassing not just technical skills and understanding but also personal attributes, competencies and qualities enabling students to become responsible members of their professional and academic fields as well as of the societies they live in. Graduates of JU will be equipped with the ability to find employment and to pursue a successful professional career, which means that graduates

will be able to:

- acquire knowledge rapidly, gather, evaluate and interpret relevant information and evaluate new concepts critically to derive scientifically founded judgements;
- apply their knowledge, understanding and methodological competences to their activity or profession to solve problems;
- present themselves and their ideas effectively and to negotiate successfully;
- demonstrate understanding and knowledge of business principles and processes and to manage projects efficiently and independently;
- take responsibility for their and their team's learning and development.

Graduates of JU will also be equipped with a foundation to become globally responsible citizens, which includes the following attributes and qualities:

- graduates have gained intercultural competence; they are aware of intercultural differences and possess skills to deal with intercultural challenges; they are familiar with the concept of tolerance;
- graduates can apply problem-solving skills to negotiate and mediate between different points of view and to manage conflicts;
- graduates can rely on basic civic knowledge; they are able to analyse global issues of economic, political, scientific, social or technological nature; they are able to evaluate situations and take decisions based on ethical considerations;
- graduates are able and prepared to take on responsibility for their professional community and society.

1.5 Career Options

Because Chemistry is the core science relating to nearly every aspect of our lives, career opportunities are diverse and abundant. Career possibilities range from research and development in the areas of pharmaceuticals, nanotechnology, materials and energy to environmental monitoring and forensic science. Chemists are also successful in other positions such as: production managers, business consultants, medical doctors, patent attorneys, marketers and even politicians!

1.6 More Information and Contact

For more information please contact the study program chairs:

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Wisdom Professor of Chemistry
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Telephone: +49 421 200-3585

and Dr. Ulrich Kortz
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Telephone: +49 421 200-3235

or visit our program website: <http://www.jacobs-university.de/chemistry-program>

2 The Curricular Structure

2.1 General

The undergraduate education at Jacobs University equips students with the key qualifications necessary for a successful academic, as well as professional career. By combining disciplinary depth and transdisciplinary breadth, supplemented by skills education and extracurricular elements, students are prepared to be responsible and successful citizens within the societies they work and live in.

The curricular structure provides multiple elements enhancing employability, transdisciplinarity, and internationality. The unique Jacobs Track, offered across all study programs, provides a broad range of tailor-made courses designed to foster career competencies. These include courses which promote communication, technology, business, (German) language, and management skills. The World Track, included in the third year of study, provides extended company internships or study abroad options. Thus students gain training on the job and intercultural experiences.

All undergraduate programs at Jacobs University are based on a coherently modularized structure, which provides students with a broad and flexible choice of study plans to meet their major as well as minor study interests.

The policies and procedures regulating undergraduate study programs at Jacobs University in general can be found on the website.

2.2 The Jacobs University 3C-Model

Jacobs University offers study programs according to the regulations of the European Higher Education Area. All study programs are structured along the European Credit Transfer System (ECTS), which facilitates credit transfer between academic institutions. The three-year undergraduate program involves six semesters of study with a total of 180 ECTS credits. The curricular structure follows an innovative and student-centered modularization scheme - the 3C-Model - which groups the disciplinary content of the three study years according to overarching themes:

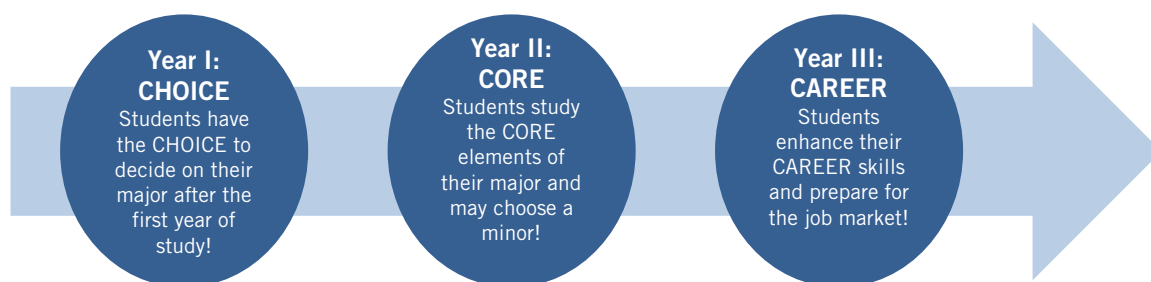


Figure 1: The Jacobs University 3C-Model

2.2.1 YEAR 1 - CHOICE

The first study year is characterized by a broad offer in disciplinary and interdisciplinary education. Students select three CHOICE modules from a variety of study programs. As a unique asset, our curricula allow students to select their study program freely from among the three selected CHOICE modules during their first year of study.

2.2.2 YEAR 2 - CORE

In the second year, students take three in-depth, discipline-specific CORE modules. One CORE module can also be taken from a second, complementary discipline, which allows students to incorporate a minor study track into their undergraduate education. Students will generally qualify for a minor if they have successfully taken at least one CHOICE module and one CORE module in a second field, and this extra qualification will be highlighted in the transcript.

2.2.3 YEAR 3 - CAREER

During their third year, students must decide on their career after graduation. In order to facilitate this decision, the fifth semester introduces two separate tracks. By default students are registered for the World Track.

1. The World Track

In this track there are two mandatory elective options:

- **Internship**

The internship program is a core element of Jacobs University's employability approach. It includes a mandatory semester-long internship off-campus (minimum 16 weeks in full-time) which provides insight into the labor market as well as practical work experience related to the respective area of study. Successful internships may initiate career opportunities for students.

As an alternative to the regular internship, a limited number of students have the opportunity to prepare in a structured manner the formation of their own start-up in the 5th semester, and can attain 20 ECTS for this study-related achievement. Jacobs University cooperates with the City Accelerator Bremen (CAB) to which students can be admitted. There are several requirements which must be fulfilled before the 5th semester in order to be admitted to the CAB, i.e. attendance of specific seminars and workshops and the successful presentation of the business idea within the framework of a competition (pitch). The module is successfully completed, when the student / team of students have submitted the business plan to CAB.

For further information, please contact the Career Services Center (<http://www.jacobs-university.de/career-services/contact>).

- **Study Abroad**

Students can take the opportunity to study abroad at one of our partner universities. Courses recognized as study abroad credits need to be pre-approved according to the Jacobs University study abroad procedures and carry minimum of 20 ECTS credits

in total. Several exchange programs allow you to be directly enrolled at prestigious partner institutions worldwide. Jacobs University's participation in Erasmus+, the European Union's exchange program, provides an exchange semester at a number of European universities including Erasmus study abroad funding.

For more information, please contact the International Office (<http://intoffice.user.jacobs-university.de/outgoing/>).

2. The Campus Track

Alternatively, students may also opt to follow the Campus Track by continuing their undergraduate education at Jacobs, namely by selecting an additional CORE module during their third year and redistributing the remaining courses and modules across the third year. This opportunity can be used by students to more intensively focus on their major or to fulfill the minor requirements for a second field of interest.

In the sixth semester, all students select from a range of specialization courses within their study program and concentrate on their Bachelor thesis in the context of a Project/Thesis Module.

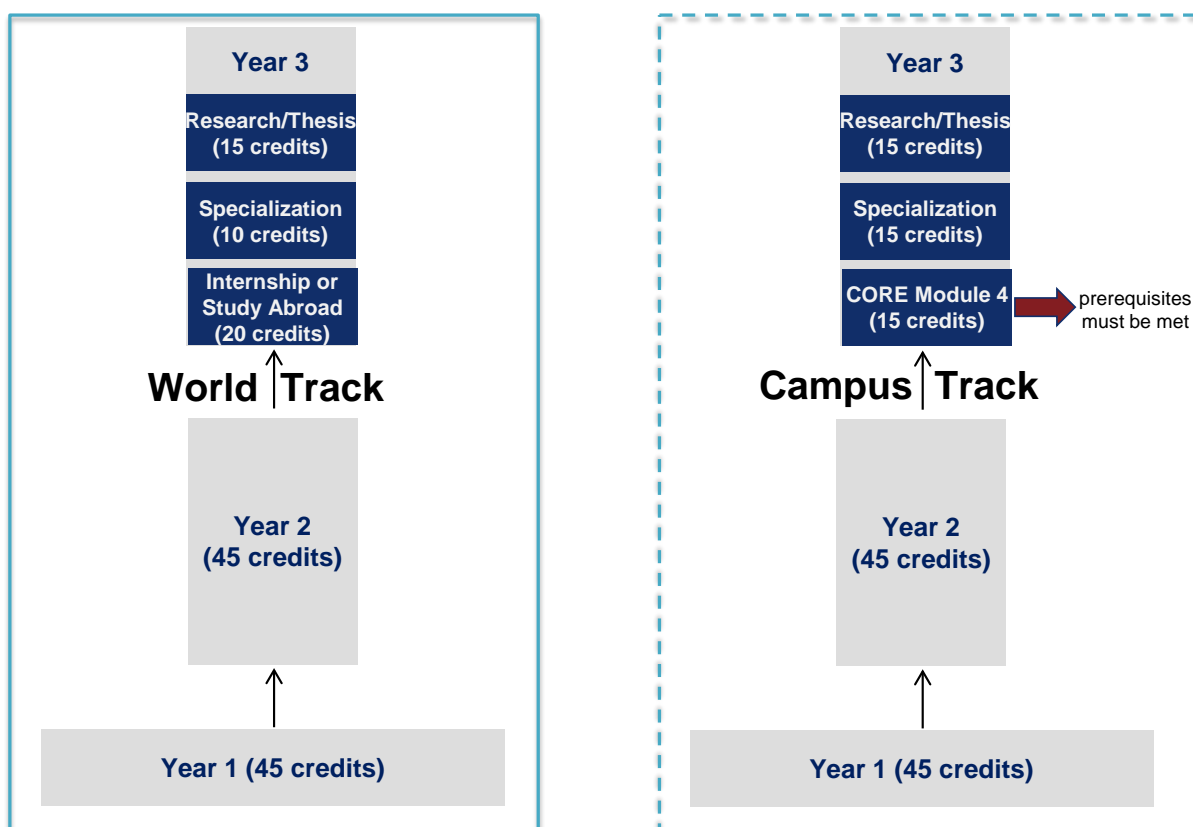


Figure 2: World Track versus Campus Track

Career Advising

Is a mandatory component of the Jacobs University's Advising and Counseling Scheme. Further components are "Academic Advising" and "Psychological Counseling and Intercultural Services". Throughout their studies all students attend a mandatory set of career skills events. The mandatory "Career Skills Advising" prepares all undergraduate students at Jacobs University for the transition from student life to working life as well as for their future career. Skills, knowledge and information which are fundamental for participation in an internship or a semester abroad will be conveyed concurrently. Essential components include information sessions, compulsory workshops on various career-relevant topics as well as participation in the annual Jacobs Career Fair.

All undergraduate students will be automatically registered for "Career Skills Advising". However, every student has to keep track of his/her individual fulfillment of requirements and has to register on CampusNet for all workshops and sessions during the official registration period at the beginning of each semester. An overview of the sequence in which events should be attended is shown in the table below.

CAREER ADVISING For Undergraduate Students matriculated Fall 2018

SEMESTER	1	2	3	4	5	6
MANDATORY BASICS	CSC-INFO Session: "CSC Services" CA01-990000		CSC-INFO Session: "World Track" CA01-990026			
MANDATORY SEMINARS	Both seminars have to be attended in your first or second semester. CSC-APPLICATION TRAINING CA01-990001 CSC-SUCCESS IN STUDIES, CAREER AND LIFE CA01-990031					
MANDATORY ELECTIVE SEMINARS (seminar program subject to availability)			Attend 2 out of several career skills seminars and workshops in your third or fourth semester, i.e. <ul style="list-style-type: none"> ▪ Research & Contacting Employers • Business Etiquette ▪ Presentation Skills • Communication Skills ▪ Grad School Application Training • Self-Management ▪ Time-Management • Decision Making • Preparing for an Interview • Introduction to Project Management ▪ Career Orientation • Working in Germany ▪ Stress Management 			
OTHER MANDATORY COMPONENTS				CSC-JACOBS CAREER FAIR in February, on campus CA01-990003		
CAREER RELATED STUDY PROGRAM COMPONENTS					INTERNSHIP (World Track) or STUDY ABROAD (World Track) or CAMPUS TRACK (exceptional)	INTERNSHIP & STUDY ABROAD EVENT

Figure 3: Career Advising

2.3 The Jacobs Track

The Jacobs Track, another stand-alone feature of Jacobs University, runs parallel to the disciplinary CHOICE, CORE, and CAREER modules across all study years and is an integral part of all study programs. It reflects our commitment to an in-depth methodological education, it fosters our transdisciplinary approach, it enhances employability, and equips students with extra skills desirable in your general field of study. Additionally, it integrates essential language courses.

Mathematics, statistics, and other methods courses are offered to all students within a comprehensive Methods Module. This module provides students with general foundations and transferable techniques which are invaluable to follow the study content not only in the study program itself but also in related fields.

The Skills Module equips students with general academic skills which are indispensable for their chosen area of study. These could be, for example, programming, data handling, presentation skills, and academic writing, scientific and experimental skills.

The transdisciplinary Triangle Module offers courses with a focus on at least one of the areas of business, technology and innovation, and societal context. The offerings comprise essential knowledge of these fields for students from other majors as well as problem-based courses that tackle global challenges from different disciplinary backgrounds. Working together with students from different disciplines and cultural backgrounds in these courses broadens the students horizon by crossing the boundaries of traditional disciplines.

Foreign languages are integrated within the Language Module. Communicative skills and foreign language competence foster students intercultural awareness and enhance their employability in a globalized and interconnected world. Jacobs University supports its students in acquiring and improving these skills by offering a variety of language courses at all proficiency levels. Emphasis is put on fostering German language skills, as they are an important prerequisite for students to learn about, explore, and eventually integrate into their host country. Hence, acquiring 10 ECTS credits in German is a requirement for all students. Students who meet the requirements of the German proficiency level (e.g. native speakers) are required to select courses in any other language program offered.

2.4 Modularization of the Chemistry Program

Year 1

Take two mandatory modules listed below and select one further CHOICE module from a different study area.

Organic Chemistry (CH03-OrgChem)

We begin by reestablishing atomic structure, and the importance of Lewis dot structures, resonance, valence-shell electron-pair repulsion, and valence-bond theory to give meaning to a covalent bond. Hybridization is then introduced to allow an accurate and predictive accounting of molecular shape. This foundation permits the introduction of: functional groups, conformation, chirality, acidity and basicity, and the basics of equilibria, thermodynamic, and kinetic phenomena. With these concepts in hand, we develop organic reactivity by examining the mechanistic pathways (arrow pushing) and chemical principles behind substitution, elimination, and addition reactions. Common reagents and functional group transformations are then learned in the context of the importance of their order and type (retrosynthetic analysis and strategy) for brevity in synthesis.

Principles of Chemistry and Physics (CH05-PrincChemPhys)

The bi-functional module Principles of Chemistry and Physics provides an introduction to basic concepts of Inorganic/General Chemistry and selected topics of Physics. Two introductory lecture courses (General Chemistry (focus on atomic structure, stoichiometry, reactions, periodic table, gases, bonding, liquids, solids) and Thermodynamics and Optics (focus on thermodynamics from physics perspective and on basic optical phenomena and instruments) are complemented by laboratory courses (General Chemistry Lab and Thermodynamics and Optics Lab) to develop fundamental practical and experimental skills.

Year 2

Take all three modules or replace one with a CORE module from a different study program.

Chemical Biotechnology (CO07-ChemBiotec)

This module provides insight into how biotechnology impacts chemical production. The replacement of both chemical catalysts by enzymes and cells and of fossil resources by renewable raw materials are two aspects which are increasingly pushed by the chemical industry in order to achieve a more sustainable production of bulk and fine chemicals, building blocks for chemical industry as well as food ingredients, bioplastics, and biofuels. Using a number of commercially successful examples as well as current R&D efforts of chemical industry, the students will be introduced into the advantages and practice of implementing cells or enzymes for the production of industrially relevant products. Moreover, the module describes the utilization of biomass and biomass waste streams as feedstock for production of the above mentioned compounds. The concept of biorefinery is also discussed.

Physical and Analytical Chemistry (CO08-PhysChem)

Teaching physical principles and measurements to understand the properties of matter. The course Physical Chemistry introduces fundamental thermodynamical principles, intermolecular forces, electrochemistry as well as underlying physical principle of chemical kinetics. The students will apply the course content to experiments on osmotic pressure, electrochemistry and

optical instrumentation. The analytical chemistry part will provide an overview over the physical principles of spectroscopic and separation methods and their application in quantitative and qualitative analysis.

Inorganic and Supramolecular Chemistry (CO09-InOrgSuMolChem)

The module gives information about inorganic materials and polymers, and about organic materials and polymers. Coordination compounds as basis of inorganic materials will be discussed. Methods for structure elucidation of polymeric and solid materials will be presented. Basic reactions to form these materials will be given. Industrially important materials and their preparation will be discussed. Examples of non-covalent interactions as basis for supramolecular chemistry are shown, and sensors based on the different technologies will be discussed. An introduction into surface and colloid chemistry forms part of the module.

Some CORE Modules require students to have taken a specific CHOICE Module. Please see the Module Handbook for details regarding pre-requisites.

Year 3

In the 3rd year students follow the World Track by default:

1. World Track

5th Semester

- Internship / study abroad

6th Semester

- Chemistry Project / Thesis Module
- Program-specific Specialization Module

Exemplary course offering:

- Advanced Synthesis
- Organometallic Chemistry
- Methods for Bioconjugation
- Structure Elucidation of Biomolecules
- Introduction to Biophysical Chemistry
- Binding and Enzyme Assays

2. Campus Track

Students who do not enter the World Track follow the Campus Track.

5th and 6th Semester

- Program-specific Project / Thesis Module
- Program-specific Specialization Module
(please see World Track for exemplary course offering)
- Additional CORE Module

2.5 The Bachelor Thesis / Project

This module is a mandatory graduation requirement for all undergraduate students. It consists of two components in the major study program guided by a Jacobs Faculty member:

1. **A Research Project** (5 ECTS)
and
2. **The Bachelor Thesis** (10 ECTS)

The workload for the project component is about 125 hours and for the thesis component about 250 hours. The title of the thesis will be shown on the transcript.

2.5.1 Aims

Within this module, students apply knowledge they have acquired about their major discipline, skills, and methods to become acquainted with actual research topics, ranging from the identification of suitable (short-term) research projects, preparatory literature searches, the realization of discipline-specific research, and the documentation, discussion, and interpretation of the results. Research results obtained from the Research Project can be embedded in the Bachelor Thesis.

2.5.2 Intended Learning Outcomes

1. Research Project

This module component consists of a guided research project in the major study program. The well-defined research task must be completed and documented according to the scientific standards in the respective discipline. It involves a high degree of independence, supported by individualized instructor feedback and guidance.

2. Bachelor Thesis

With their Bachelor Thesis students should demonstrate mastery of the contents and methods of the major specific research field. Furthermore, students should show the ability to analyze and solve a well-defined problem with scientific approaches, a critical reflection of the status quo in scientific literature, and an original development of their own ideas.

Both, the Research Project and the Bachelor Thesis, can also have an inter- or transdisciplinary nature - with the explicit permission of the supervisor.

2.5.3 Supervision

Both module components can be performed with the same Jacobs faculty member, or different ones, the latter in order to allow a broader research experience. Students are required to choose a supervisor, at the latest, by the end of the drop-add period of the semester in which the module component is taken. **The selected supervisor(s) must approve the Project topic and Bachelor Thesis topic before the student starts to work towards the module component.** The respective study program chairs will assist in the search for prospective supervisor(s).

2.5.4 Registration

World Track students register for both components, at the earliest, in their 6th semester.

Campus Track students register for the Project component in the 5th and for the Bachelor Thesis component, at the earliest, in their 6th semester.

The registrations must be made before the end of the respective drop/add periods.

Later enrolment is possible for those students pursuing a second major or those who graduate late for other reasons. These students perform their (second) thesis earliest in the 7th semester of their studies. They have to contact the Student Records Office for individual registration.

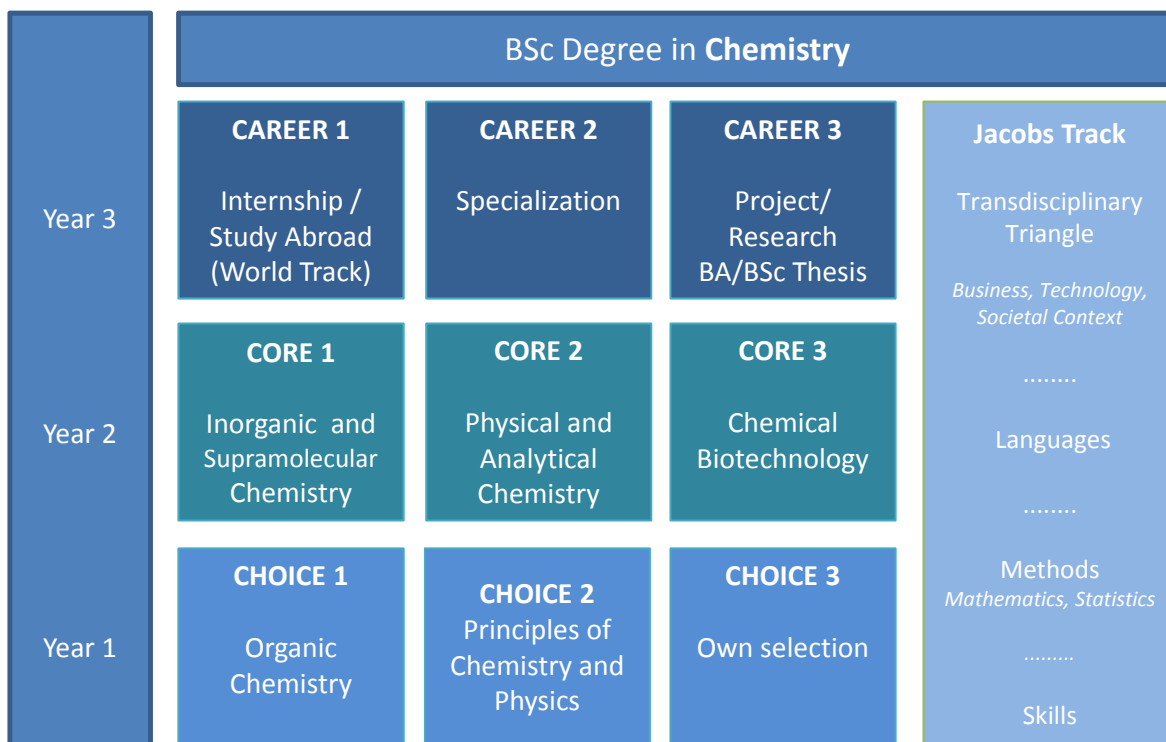
Students are allowed to extend their thesis related work into the intersession or summer break upon approval of the thesis supervisor and Student Records. Students are not allowed to register for different Bachelor Thesis courses in the same semester.

2.5.5 Formal Regulations for the Bachelor Thesis

- **Timing**
The Thesis work has to be generated within the semester of registration. The semester period has 14 weeks.
- **Extent**
The document must be between 15-25 pages in length, including references, but excluding appendices or supporting information. Deviations in length and format can be determined within individual study programs and should be communicated to all registered students by the study program chair.
- **Cover page**
The cover page must show the title of the Bachelor Thesis, the university's name, the month and year of submission, the name of the student and the name of the supervisor.
- **Statutory Declaration**
Each Bachelor Thesis must include a statutory declaration signed by the student confirming it is their own independent work and that it has not been submitted elsewhere. The respective form can be found on the Student Records Office website.
- **Submission**
The Bachelor Thesis must be submitted as a hard copy (pdf-file) to the supervisor and additionally to the Student Records Office via online form on the Student Records Office website.

Deadline for submission of the Bachelor Thesis is May 15 (unless specified otherwise by the Student Records Office).

2.6 Structure



YEAR 1 Take three CHOICE modules, one free selection

YEAR 2 Take three CORE modules, one CORE module can be substituted by a CORE module from a second study program to pursue a minor

YEAR 3 Alternatively Campus Track with a 4th CORE module instead of internship/study abroad module

Figure 4: Chemistry Module Structure

3 Appendix 1a/1b: Mandatory Module and Examination Plans for World Track and Campus Track

Jacobs University Bremen reserves the right to substitute courses by replacements and/or reduce the number of mandatory/mandatory elective courses offered.

4 Appendix 2: Course Data for Program-Specific CHOICE and CORE Courses

All course data stated in the appendix is based on the previous study year and subject to change.

Appendix 2 - Course Data

Course Name Organic Chemistry I	Course No CH03-400102	ECTS 5
Module Affiliation CH03-OrgChem Organic Chemistry	Workload (hrs / sem) Contact Time: 35,00 Private Study: 90,00	Level Bachelor 1st Year CHOICE
Course Description / Content / Aims The course begins by establishing a strong working knowledge of atomic, hybridized, and molecular orbitals. This is vital for our understanding of the next material: Lewis dot structures, octet rule, electron ownership, resonance, bond angles, bond strength, bond order, molecular shape, conformation, transition states, and pKa (acidity/basicity). Functional groups (halides, alcohols, ethers, amines, aldehydes, ketones, carboxylic acids, esters, amides, anhydrides, acetals, etc.), nomenclature, and chirality (stereogenic centers) are then addressed. This conceptual framework allows reactivity (electrophiles and nucleophiles) to then be discussed in the context of mechanistic pathways (SN1, E1, SN2, and E2) and the law of mass action. Specific functional group interconversions are then addressed beginning with an early discussion of carbonyl reactivity (enolates, imines, enamines, aldol reactions, etc.) to better facilitate our understanding of biochemical reaction pathways.		
Methods of Assessment		
Name		Weighting
Final Exam		40%
Midterm Exam		36%
Quizz(es)		24%
Course Name Organic Chemistry II	Course No CH03-400103	ECTS 5
Module Affiliation CH03-OrgChem Organic Chemistry	Workload (hrs / sem) Contact Time: 35,00 Private Study: 90,00	Level Bachelor 1st Year CHOICE
Course Description / Content / Aims The second semester of Organic Chemistry strongly builds on the concepts and principles introduced during the first semester. During this course a much broader exposure to reagents and reactions (functional group interconversion) is focused on, and this material is learned within the framework of stereocontrol, mechanisms (arrow pushing), and the importance of reaction step order to achieve step efficient synthesis. Exposure to the tactics and strategies of synthesis will provide the context for retrosynthetic analysis, and an appreciation synthesizing challenging drug molecules. Major topics of discussion will be: alkene formation (Wittig reaction) and transformations thereof (bromination, epoxidation, dihydroxylation, Diels-Alder reactions, etc.), aromaticity, Friedel-Crafts alkylation and acylation, benzyne elimination-addition reactions, derivatization and formation of phenols (Meisenheimer complexes - S _N Ar mechanism), a continued investigation of carbonyl chemistry (aldehyde, ketone, ester, amides, carboxylic acid): formation, reduction, nucleophilic addition to, etc. Amines play a vital role in drug development and their formation and manipulation are discussed. Special topics will be introduced, e.g. amide hydrolysis (peptide cleavage) at an oxyanion hole, to show how a living system can accomplish important chemical reactions.		
Methods of Assessment		
Name		Weighting
Final Exam		40%
Midterm Exam 1		25%
Midterm Exam 2		35%

Appendix 2 - Course Data

Course Name Organic Chemistry I Lab	Course No CH03-400112	ECTS 2,5
Module Affiliation CH03-OrgChem Organic Chemistry	Workload (hrs / sem) Contact Time: 25,50 Private Study: 37,00	Level Bachelor 1st Year CHOICE
Course Description / Content / Aims A chemical laboratory is a place for exploration, but before that happens we must reinforce important safety aspects, common hazards, and the structure & content requirements of a useful laboratory report. After this, we introduce the essential techniques to: monitor and quench reactions (TLC, color change, neutralizing active chemicals, etc.), purify products (chromatography, crystallization, separatory funnel extractions, etc.), and spectroscopically identify compounds. In parallel, we introduce the equipment (rotary evaporator, melting point apparatus, etc.) and instrumentation (nuclear magnetic resonance (1H and 13C NMR)) to achieve those goals. After mastering these techniques, the next semester (Org Chem II lab) is devoted to the higher level goal of setting up reactions under diverse reaction conditions to produce pure compounds.		
Methods of Assessment		
Name		Weighting
Compound Synthesis		30%
General Laboratory Performance		40%
Lab Reports		30%
Course Name Organic Chemistry II Lab	Course No CH03-400113	ECTS 2,5
Module Affiliation CH03-OrgChem Organic Chemistry	Workload (hrs / sem) Contact Time: 25,50 Private Study: 37,00	Level Bachelor 1st Year CHOICE
Course Description / Content / Aims A chemical laboratory is a place for exploration, and the second semester organic laboratory places you squarely in that environment. Here you will set up your own reactions, sometimes at low temperature (e.g. -78 oC) and at other times under an inert atmosphere of nitrogen gas to protect your reactions from the negative effects of the moisture present in the air that we breathe. You will also expand your techniques, e.g., employing vacuum distillation, and exposure to instrumentation, e.g., gas and liquid chromatography, infrared spectroscopy (IR), etc. Most importantly, you will begin to appreciate the whole process of designing and then performing a reaction. From the starting reaction table you have built using reaction stoichiometry to determine the weight or volume of the reagents, to the order and timing of reagent additions, all the way to providing a pure chemical in the end whose structure you can rigorously support via multiple pieces of chromatographic and spectroscopic evidence. You will leave this lab having obtained the essence of the art of organic synthesis.		
Methods of Assessment		
Name		Weighting
General laboratory performance		40%
Laboratory reports		30%
Synthesized compounds		30%

Appendix 2 - Course Data

Course Name Thermodynamics and Optics	Course No CH05-200102	ECTS 5								
Module Affiliation CH05-PrincChemPhys Principles of Chemistry and Physics	Workload (hrs / sem) Contact Time: 35,00 Private Study: 90,00	Level Bachelor 1st Year CHOICE								
<p>Course Description / Content / Aims</p> <p>Optics and thermodynamics are topics of interest for any student enrolled in a natural science major such as physics or chemistry. This course introduces the basic physical principles and concepts of optics and thermodynamics. Emphasis is laid on general principles and phenomena necessary to describe natural phenomena and basic scientific instruments. Experiments and demonstrations are included in the lecture, and a tutorial is offered to discuss homework and topics of interest in more details.</p> <p>The course is divided into two sections: one is thermodynamics starting with the description of temperature, heat, ideal gases and kinetic gas theory; it includes the 1st law and heat engines together with the 2nd law and entropy and other thermodynamic potentials. The other section starts with an introduction to the nature of light and waves, and the basic phenomena of reflection and refraction. It then focuses on lenses and optical instruments before it turns to interference and diffraction and ends with advanced topics such as spectroscopy and lasers.</p>										
<p>Methods of Assessment</p> <table> <thead> <tr> <th>Name</th> <th>Weighting</th> </tr> </thead> <tbody> <tr> <td>First Exam</td> <td>40%</td> </tr> <tr> <td>Home Work</td> <td>20%</td> </tr> <tr> <td>Second Exam</td> <td>40%</td> </tr> </tbody> </table>			Name	Weighting	First Exam	40%	Home Work	20%	Second Exam	40%
Name	Weighting									
First Exam	40%									
Home Work	20%									
Second Exam	40%									
Course Name Thermodynamics and Optics Lab	Course No CH05-200112	ECTS 2,5								
Module Affiliation CH05-PrincChemPhys Principles of Chemistry and Physics	Workload (hrs / sem) Contact Time: 60,00 Private Study: 120,00	Level Bachelor 1st Year CHOICE								
<p>Course Description / Content / Aims</p> <p>Physics is an experimental science and the ultimate test of any theory or description of nature is the experiment. This lab course complements the Thermodynamics and Optics lecture with experiments in those fields and is supplemented by experiments covering physics topics from the other lecture of the module, inorganic chemistry. It deepens the understanding and extends the topics covered in the lecture, which is a co-requisite for this course. Prior to the course, students need to attend the relevant safety instructions and will get an introduction to error analysis. The lab offers six different experiments and runs over six afternoons.</p> <p>The aim of the lab sessions is hands-on experience on how to investigate physical phenomena and topics presented in the lecture; to plan, carry out, and analyze experiments in physics; to describe, summarize and present experimental results adequately. Examples of experiments include Coulomb force, ideal gas and diffraction.</p>										
<p>Methods of Assessment</p> <table> <thead> <tr> <th>Name</th> <th>Weighting</th> </tr> </thead> <tbody> <tr> <td>Final Exam</td> <td>34%</td> </tr> <tr> <td>Reports</td> <td>66%</td> </tr> </tbody> </table>			Name	Weighting	Final Exam	34%	Reports	66%		
Name	Weighting									
Final Exam	34%									
Reports	66%									

Appendix 2 - Course Data



Course Name General Chemistry	Course No CH05-400101	ECTS 5
Module Affiliation CH05-PrincChemPhys Principles of Chemistry and Physics	Workload (hrs / sem) Contact Time: H 1€€ Private Study: J 6€€	Level Bachelor 1st Year CHOICE
Course Description / Content / Aims An introduction to inorganic and general chemistry covering the following areas: (1) Chemical Foundations, (2) Atoms, Molecules, and Ions, (3) Stoichiometry, (4) Types of Chemical Reactions and Solution Stoichiometry, (5) Gases, (6) Atomic Structure and Periodicity, (7) Bonding: General Concepts, (8) Covalent Bonding: Orbitals, (9) Liquids and Solids, (10) Properties of Solutions		
Methods of Assessment		
Name		Weighting
Attendance		10%
Final Exam		45%
Midterm Exam		45%
Course Name General Chemistry Lab	Course No CH05-400111	ECTS 2,5
Module Affiliation CH05-PrincChemPhys Principles of Chemistry and Physics	Workload (hrs / sem) Contact Time: G 1€€ Private Study: H 1€€	Level Bachelor 1st Year CHOICE
Course Description / Content / Aims Foundation principles of chemistry, including basic laboratory techniques, qualitative analysis of anions and cations, strong/weak acids and bases, titrations, solubility of salts, crystallization, redox reactions, gravimetric analysis, volumetric analysis, complex formation, synthesis of nanoparticles.		
Methods of Assessment		
Name		Weighting
Performance		50%
Quizz(es)		10%
Reports		40%

Appendix 2 - Course Data

Course Name Enzymes and Cells in Biochemical Production	Course No CO07-400104	ECTS 5
Module Affiliation CO07-ChemBiotec Chemical Biotechnology	Workload (hrs / sem) Contact Time: 35,00 Private Study: 90,00	Level Bachelor 2nd Year CORE
Course Description / Content / Aims This lecture gives an overview of how cells and enzymes contribute to the synthesis of chemical compounds. Examples will demonstrate that biological systems can be utilized in a cost-effective manner for the mild, environmentally friendly production of chemicals. Biological conversions include biotransformations using isolated enzymes and whole cells for specific chemical conversion steps (to replace chemical catalysts) as well as de novo synthesis of complex chemicals or their precursors from simple nutrients by living cells. The basic principles of cellular metabolism will be discussed in order to provide a fundamental knowledge of how it can be exploited for chemical production. The students will read and present state-of-the-art literature in the field.		
Methods of Assessment		
Name		Weighting
Active Participation		10%
Final Exam		60%
Presentation		30%
Course Name Bioproducts and Biosystems Engineering	Course No CO07-400105	ECTS 5
Module Affiliation CO07-ChemBiotec Chemical Biotechnology	Workload (hrs / sem) Contact Time: 35,00 Private Study: 90,00	Level Bachelor 2nd Year CORE
Course Description / Content / Aims This lecture explores the utilization of biomass as a feedstock for the production of biofuels, block chemicals and fine biochemical. Biomass conversion routes are described, namely the sugar & cellulose platform and the thermochemical platform e.g. gasification and pyrolysis. The biological and engineering aspects of such bio-refinery approaches are discussed. The design and implementation of separation methods (downstream processing) is also presented. The implementation of a bio-refinery operation is analyzed in the context of distributed biomass conversion systems or centralized plants. Emphasis is given on the dovetailing of thermochemical and fermentation-based processes with downstream recovery and purification schemes from the laboratory to the pilot plant and beyond.		
Methods of Assessment		
Name		Weighting
Final Exam		100%

Appendix 2 - Course Data

Course Name Biochemical Production Lab	Course No CO07-400114	ECTS 2,5
Module Affiliation CO07-ChemBiotec Chemical Biotechnology	Workload (hrs / sem) Contact Time: 25,50 Private Study: 37,00	Level Bachelor 2nd Year CORE
Course Description / Content / Aims The lab course will provide the students with basic skills of how to practically work with enzymes and microbial cells in order to produce compounds relevant in chemical industry (biofuels, bulk chemicals and fine chemicals). A major focus will be put on the use of renewable resources for chemical production. The students will learn how to prepare media, run microbial fermentations and analyze product formation and other important process parameters.		
Methods of Assessment		
Name		Weighting
Final Report		100%
Course Name Bioproducts and Biosystems Engineering Lab	Course No CO07-400115	ECTS 2,5
Module Affiliation CO07-ChemBiotec Chemical Biotechnology	Workload (hrs / sem) Contact Time: 25,50 Private Study: 37,00	Level Bachelor 2nd Year CORE
Course Description / Content / Aims This Laboratory demonstrates the utilization of biomass as a feedstock for the production of biofuels, block chemicals and fine biochemical. The design and implementation of separation methods (downstream processing) is presented in a set of related experiments e.g. the determination of dynamic and static binding capacities of adsorbent materials, the performance of membrane separation technologies, and the utilisation of fractionation routines based on selective precipitation. Emphasis is given on the dovetailing of upstream and downstream recovery and purification schemes from the laboratory to the pilot plant and beyond.		
Methods of Assessment		
Name		Weighting
Final Report		100%

Appendix 2 - Course Data

Course Name Analytical Chemistry	Course No CO08-400121	ECTS 5
Module Affiliation CO08-PhysChem Physical and Analytical Chemistry	Workload (hrs / sem) Contact Time: 35,00 Private Study: 90,00	Level Bachelor 2nd Year CORE
Course Description / Content / Aims Analytical chemistry is an important applied area of chemistry. Analytical chemistry deals with the separation, identification and quantification of any chemical compound. It therefore provides an interface between the traditional areas of organic, inorganic and physical chemistry with life sciences and all other areas of science requiring identification and quantification of chemical compounds. Analytical chemistry provides the tools for all areas of experimental chemistry and a good foundation of analytical techniques is not only expected of any chemist but also life scientist. The course will give an introduction into analytical chemistry with selected applications in environmental chemistry. This will include an introduction to analytical terms and definitions, basic statistic treatment of experimental data, qualitative and quantitative analysis and instrumental analysis with emphasis on spectroscopic techniques such as UV-Vis, NMR, mass spectrometry, IR and Raman spectroscopy and fluorimetry. Furthermore separation techniques such as HPLC and GC will be covered.		
Methods of Assessment		
Name	Weighting	
Final Exam	60%	
Home Work	20%	
Quizz(es)	20%	
Course Name Physical Chemistry	Course No CO08-400211	ECTS 5
Module Affiliation CO08-PhysChem Physical and Analytical Chemistry	Workload (hrs / sem) Contact Time: 35,00 Private Study: 90,00	Level Bachelor 2nd Year CORE
Course Description / Content / Aims This course covers classical chemical thermodynamics and kinetics. The thermodynamics part includes the laws of thermodynamics, energy, enthalpy, entropy and free energy. They are applied to physical processes as well as chemical reactions. The kinetics part treats basic chemical kinetics laws, reactions of different order and applications to complex reaction sequences with the appropriate approximation. A final part deals with the kinetics of diffusion and ion mobility in fluid mixtures.		

Appendix 2 - Course Data

Course Name Analytical Chemistry Lab	Course No CO08-400231	ECTS 2,5
Module Affiliation CO08-PhysChem Physical and Analytical Chemistry	Workload (hrs / sem) Contact Time: 25,50 Private Study: 37,00	Level Bachelor 2nd Year CORE
Course Description / Content / Aims Analytical chemistry is an important applied area of chemistry. The Analytical chemistry laboratory will give the students practical experience in the use of modern analytical instrumentation used for compound identification and structure elucidation. The laboratory sessions will provide set experiments using IR-, UV-VIS-NMR-spectroscopy, mass spectrometry and chromatographic techniques including gas chromatography and HPLC. Within the course students will as well learn to critically interpret experimental data and apply basic statistical operations and error calculus to experimental data.		
Methods of Assessment		
Name		Weighting
Tests / Reports		100%
Course Name Physical Chemistry Lab	Course No CO08-400262	ECTS 2,5
Module Affiliation CO08-PhysChem Physical and Analytical Chemistry	Workload (hrs / sem) Contact Time: 25,50 Private Study: 37,00	Level Bachelor 2nd Year CORE
Course Description / Content / Aims This course provides training in the experimental background of physical chemistry. Examples to be covered are: Microcalorimetry, Particle sizing, Zetapotential, UV-vis and fluorescence as a tool for binding, Reaction enthalpy, Phase transition, Osmotic Pressure, Electrochemistry,		

Appendix 2 - Course Data

Course Name Inorganic Chemistry	Course No CO09-400221	ECTS 5
Module Affiliation CO09-InorgSuMolChem Inorganic and Supramolecular Chemistry	Workload (hrs / sem) Contact Time: 35,00 Private Study: 90,00	Level Bachelor 2nd Year CORE
Course Description / Content / Aims This course introduces advanced concepts of inorganic chemistry, such as Molecular Structure and Bonding (VB theory, MO theory, semiconductors), Symmetry and Group Theory, Structures of Solids (metals, ionic solids), d-metal Complexes (structure and symmetry, bonding and electronic structure, reactions of complexes), The Electronic Spectra of Complexes (electronic spectra of atoms vs complexes, bonding and spectra of M-M bonded compounds).		
Methods of Assessment		
Name	Weighting	
Active Participation	20%	
Final Exam	40%	
Midterm Exam	40%	
Course Name Inorganic Chemistry Lab	Course No CO09-400232	ECTS 2,5
Module Affiliation CO09-InorgSuMolChem Inorganic and Supramolecular Chemistry	Workload (hrs / sem) Contact Time: 25,50 Private Study: 37,00	Level Bachelor 2nd Year CORE
Course Description / Content / Aims Synthesis, separation, purification and characterization of inorganic main-group and transition metal compounds. Quantitative analysis (gravimetric and spectrometric). Kinetics of inorganic reactions. Instrumentation used: FT-IR, UV-vis, AA, TGA, XRD, and NMR.		
Methods of Assessment		
Name	Weighting	
Lab Performance	40%	
Lab Reports	50%	
Quizz(es)	10%	

Appendix 2 - Course Data

Course Name Supramolecular Chemistry	Course No CO09-420432	ECTS 5
Module Affiliation CO09-InorgSuMolChem Inorganic and Supramolecular Chemistry	Workload (hrs / sem) Contact Time: 35,00 Private Study: 90,00	Level Bachelor 2nd Year CORE
Course Description / Content / Aims The course will introduce the principal design of nanomolecular systems, their characteristic properties, principal functions and applications. The focus will lie on organic and biological functional systems and on applications in the physical sciences, thus complementing the inorganic, materials-related and methodological aspects covered in Nanomolecular Science. It deals with supramolecular chemistry, the construction of nanomolecular functional systems including shuttles, photonic devices, molecular rotors and machines, and molecular containers, including biological examples and applications, e.g., drug delivery and highthroughput screening. Synthetic aspects of supramolecular assemblies, both of discrete monodisperse nature (calixarenes, functionalized fullerenes, cyclodextrins, cucurbiturils), as well as less well defined polydisperse systems like dendrimers will be discussed.		
Methods of Assessment		
Name		Weighting
Final Exam		60%
Test / Report		40%
Course Name Supramolecular Chemistry Lab	Course No CO09-420434	ECTS 2,5
Module Affiliation CO09-InorgSuMolChem Inorganic and Supramolecular Chemistry	Workload (hrs / sem) Contact Time: 25,50 Private Study: 37,00	Level Bachelor 2nd Year CORE
Course Description / Content / Aims The goal of this lab course is to familiarize students with supramolecular chemistry in the form of hands-on lab experiences. Topics covered will include experiments concerning the formation and self-assembly of supramolecular, nanoscale structures, the thermodynamics, kinetics, and concentration-dependent phenomena of supramolecular interactions and supramolecular photochemistry. The students will synthesize supramolecular structures, perform titration experiments and analyze the obtained data. The lab course will provide a deeper understanding of receptor-ligand binding and the factors that influence binding strength and efficiency. A key part will also be the design of meaningful control experiments to exclude alternative explanations for the observed effects.		
Methods of Assessment		
Name		Weighting
3 Lab Reports		60%
3 Oral Entrance Exams		40%

Appendix 1a - Mandatory Module and Examination Plan for World Track



Chemistry – World Track

Matriculation Fall 2018

Program-Specific Modules	Type	Status ¹	Semester	Credits	Jacobs Track Modules (General Education)	Type	Status ¹	Semester	Credits			
Year 1 - CHOICE					45						20	
<i>Take the two mandatory CHOICE modules listed below, these are a requirement for the Chemistry program.</i>												
CH03-OrgChem	Module: Organic Chemistry			m	15	JT-ME-MethodsMath	Module: Methods / Mathematics			m	7,5	
CH03-400102	Organic Chemistry I	Lecture	m	1	5	JT-ME-120106 / 120103	Take either Applied Calculus I or Calculus I	Lecture	m	1	2,5	
CH03-400112	Organic Chemistry I Lab	Lab	m	1	2,5	JT-ME-120107 / 120104	Take either Applied Calculus II or Calculus II	Lecture	m	1	2,5	
CH03-400103	Organic Chemistry II	Lecture	m	2	5	JT-ME-120101	Mathematical Concepts in the Sciences	Lecture	m	2	2,5	
CH03-400113	Organic Chemistry II Lab	Lab	m	2	2,5	JT-SK-Skills	Module: Skills			m	2,5	
CH05-PrincChemPhys	Module: Principles of Chemistry and Physics			m	15	JT-SK-990103	Scientific and Experimental Skills	Lecture	m	1	2,5	
CH05-400101	General Chemistry	Lecture	m	1	5	JT-TA-TriArea	Module: Triangle Area			m	5	
CH05-400111	General Chemistry Lab	Lab	m	1	2,5		Take two courses from the triangle (BUSINESS, TECHNOLOGY & INNOVATION, SOCIETAL CONTEXT) area. Each counts 2,5 ECTS ³		me	1/2	5	
CH05-200102	Thermodynamics and Optics	Lecture	m	2	5	JT-LA-Language	Module: Language			m	5	
CH05-200112	Thermodynamics and Optics Lab	Lab	m	2	2,5		Take two German courses (2,5 ECTS each). Native German speakers take courses in another offered language	Seminar	me	1/2	5	
	Module: CHOICE (own selection)			e	1/2	15	CA01-CarAdv	Career Advising⁴				
Students take one further CHOICE module from those offered for all other study programs. ²												
Year 2 - CORE					45						20	
<i>Take all three modules <u>or</u> replace one with a CORE module from a different study program. ²</i>												
CO09-InorgSuMolChc	Module: Inorganic and Supramolecular Chemistry			me	15	JT-ME-MethodsMath	Module: Methods / Mathematics			m	7,5	
CO09-420432	Supramolecular Chemistry	Lecture	m	3	5	JT-ME-400205	Methods in Analytical Science	Lecture/L m		3	2,5	
CO09-420434	Supramolecular Chemistry Lab	Lab	m	4	2,5	JT-ME-120122	Foundations of Linear Algebra I	Lecture	m	4	2,5	
CO09-400221	Inorganic Chemistry	Lecture	m	4	5		Take one further Methods (mandatory) elective course. (2,5 ECTS). ²	Lecture	me	3/4	2,5	
CO09-400232	Inorganic Chemistry Lab	Lab	m	4	2,5	JT-TA-TriArea	Module: Triangle Area			m	7,5	
CO08-PhysChem	Module: Physical and Analytical Chemistry			me	15		Take three courses from the triangle (BUSINESS, TECHNOLOGY & INNOVATION, SOCIETAL CONTEXT) area. Each counts 2,5 ECTS ³		me	3/4	7,5	
CO08-400121	Analytical Chemistry	Lecture	m	4	5	JT-LA-Language	Module: Language			m	5	
CO08-400231	Analytical Chemistry Lab	Lab	m	4	2,5		Take two German courses (2,5 ECTS each). Native German speakers take courses in another offered language	Seminar	me	3/4	5	
CO08-400211	Physical Chemistry	Lecture	m	3	5	CA01-CarAdv	Career Advising⁴					
CO08-400262	Physical Chemistry Lab	Lab	m	4	2,5							
CO07-ChemBiotec	Module: Chemical Biotechnology			me	15							
CO07-400104	Enzymes and Cells in Biochemical Production	Lecture	m	3	5							
CO07-400114	Biochemical Production Lab	Lab	m	3	2,5							
CO07-400105	Bioproducts and Biosystems Engineering	Lecture	m	4	5							
CO07-400115	Bioproducts and Biosystems Engineering Lab	Lab	m	3	2,5							
Year 3 - CAREER					45						5	
CA02 / CA03	Module: Internship / Study Abroad			m	5	20	JT-SK-Skills	Module: Skills			m	2,5
							JT-SK-990104	Advanced Scientific and Experimental Skills	Lecture	m	6	2,5
CA06-CHEM	Module: Project/Thesis CHEM			m	15		JT-TA-TriArea	Module: Triangle Area			m	2,5
CA06-400303	Project CHEM		m	6	5			Take one course from the triangle (BUSINESS, TECHNOLOGY & INNOVATION, SOCIETAL CONTEXT) area. Each counts 2,5 ECTS ³		me	6	2,5
CA06-400304	Thesis CHEM		m	6	10		CA01-CarAdv	Career Advising⁴				
CAS-WT-CHEM	Module: Specialization Area CHEM			m	10							
	Take 10 ECTS of specialization courses ²			me	5/6	10						
Total ECTS										180		

¹ Status (m = mandatory, e = elective, me = mandatory elective) ² For a full listing of all CHOICE / CORE / CAREER / Jacobs Track modules please consult the CampusNet online catalogue and / or the module handbook (on our website).

³ You are required to take six Triangle Area courses in total. Select two from each of the three triangle areas (BUSINESS, TECHNOLOGY & INNOVATION, SOCIETAL CONTEXT).

⁴ Mandatory component of the Jacobs University's Counseling and Advising Scheme.

Appendix 1b - Mandatory Module and Examination Plan for Campus Track

Chemistry – Campus Track

Matriculation Fall 2018

Program-Specific Modules	Type	Status ¹	Semester	Credits	Jacobs Track Modules (General Education)	Type	Status ¹	Semester	Credits				
Year 1 - CHOICE					45						20		
<i>Take the two mandatory CHOICE modules listed below, these are a requirement for the Chemistry program.</i>													
CH03-OrgChem	Module: Organic Chemistry			m	15	JT-ME-MethodsMath	Module: Methods / Mathematics			m	7,5		
CH03-400102	Organic Chemistry I	Lecture	m	1	5	JT-ME-120106 / 120103	Take either Applied Calculus I or Calculus I	Lecture	m	1	2,5		
CH03-400112	Organic Chemistry I Lab	Lab	m	1	2,5	JT-ME-120107 / 120104	Take either Applied Calculus II or Calculus II	Lecture	m	1	2,5		
CH03-400103	Organic Chemistry II	Lecture	m	2	5	JT-ME-120101	Mathematical Concepts in the Sciences	Lecture	m	2	2,5		
CH03-400113	Organic Chemistry II Lab	Lab	m	2	2,5	JT-SK-Skills	Module: Skills			m	2,5		
CH05-PrincChemPhys	Module: Principles of Chemistry and Physics				m	15	JT-SK-990103	Scientific and Experimental Skills	Lecture	m	1	2,5	
CH05-400101	General Chemistry	Lecture	m	1	5	JT-TA-TriArea	Module: Triangle Area			m	5		
CH05-400111	General Chemistry Lab	Lab	m	1	2,5		Take two courses from the triangle (BUSINESS, TECHNOLOGY & INNOVATION, SOCIETAL CONTEXT) area. Each counts 2,5 ECTS ³	me	1/2	5			
CH05-200102	Thermodynamics and Optics	Lecture	m	2	5	JT-LA-Language	Module: Language			m	5		
CH05-200112	Thermodynamics and Optics Lab	Lab	m	2	2,5		Take two German courses (2,5 ECTS each). Native German speakers take courses in another offered language	Seminar	me	1/2	5		
	Module: CHOICE (own selection)				e	1/2	15						
Students take one further CHOICE module from those offered for all other study programs. ²													
Year 2 - CORE					45						20		
<i>Take all three modules <u>or</u> replace one with a CORE module from a different study program. ²</i>													
CO09-InorgSuMolChem	Module: Inorganic and Supramolecular Chemistry				me	15	JT-ME-MethodsMath	Module: Methods / Mathematics			m	7,5	
CO09-420432	Supramolecular Chemistry	Lecture	m	3	5	JT-ME-400205	Methods in Analytical Science	Lecture/L m	m	3	2,5		
CO09-420434	Supramolecular Chemistry Lab	Lab	m	4	2,5	JT-ME-120122	Foundations of Linear Algebra I	Lecture	m	4	2,5		
CO09-400221	Inorganic Chemistry	Lecture	m	4	5		Take one further Methods (mandatory) elective course. (2,5 ECTS). ²	Lecture	me	3/4	2,5		
CO09-400232	Inorganic Chemistry Lab	Lab	m	4	2,5	JT-TA-TriArea	Module: Triangle Area			m	7,5		
CO08-PhysChem	Module: Physical and Analytical Chemistry				me	15		Take three courses from the triangle (BUSINESS, TECHNOLOGY & INNOVATION, SOCIETAL CONTEXT) area. Each counts 2,5 ECTS ³	me	3/4	7,5		
CO08-400121	Analytical Chemistry	Lecture	m	4	5	JT-LA-Language	Module: Language			m	5		
CO08-400231	Analytical Chemistry Lab	Lab	m	4	2,5		Take two German courses (2,5 ECTS each). Native German speakers take courses in another offered language	Seminar	me	3/4	5		
CO08-400211	Physical Chemistry	Lecture	m	3	5	CA01-CarAdv	Career Advising⁴						
CO08-400262	Physical Chemistry Lab	Lab	m	4	2,5								
CO07-ChemBiotec	Module: Chemical Biotechnology				me	15							
CO07-400104	Enzymes and Cells in Biochemical Production	Lecture	m	3	5								
CO07-400114	Biochemical Production Lab	Lab	m	3	2,5								
CO07-400105	Bioproducts and Biosystems Engineering	Lecture	m	4	5								
CO07-400115	Bioproducts and Biosystems Engineering Lab	Lab	m	3	2,5								
Year 3 - CAREER					45						5		
COXX	Module: Additional (4th) CORE module				m	5/6	15	JT-SK-Skills	Module: Skills			m	2,5
								JT-SK-990104	Advanced Scientific and Experimental Skills	Lecture	m	6	2,5
CA06-CHEM	Module: Project/Thesis CHEM				m	15	JT-TA-TriArea	Module: Triangle Area			m	2,5	
CA06-400303	Project CHEM		m	5	5		Take one course from the triangle (BUSINESS, TECHNOLOGY & INNOVATION, SOCIETAL CONTEXT) area. Each counts 2,5 ECTS ³	me	5	2,5			
CA06-400304	Thesis CHEM		m	6	10	CA01-CarAdv	Career Advising⁴						
CAS-CT-CHEM	Module: Specialization Area CHEM				m	15							
	Take 15 ECTS of specialization courses ²		me	5/6	15								
Total ECTS										180			

¹ Status (m = mandatory, e = elective, me = mandatory elective)

² For a full listing of all CHOICE / CORE / CAREER / Jacobs Track modules please consult the **CampusNet online catalogue** and / or the module handbook (on our website).

³ You are required to take six Triangle Area courses in total. Select two from each of the three triangle areas (BUSINESS, TECHNOLOGY & INNOVATION, SOCIETAL CONTEXT).

⁴ Mandatory component of the Jacobs University's Counseling and Advising Scheme.